Biosynthesis of Vitamin B₁₂: Does 5¹-Norcobyrinic Acid or 15¹-Norcobyrinic Acid Act as a Precursor of Cobyrinic Acid?

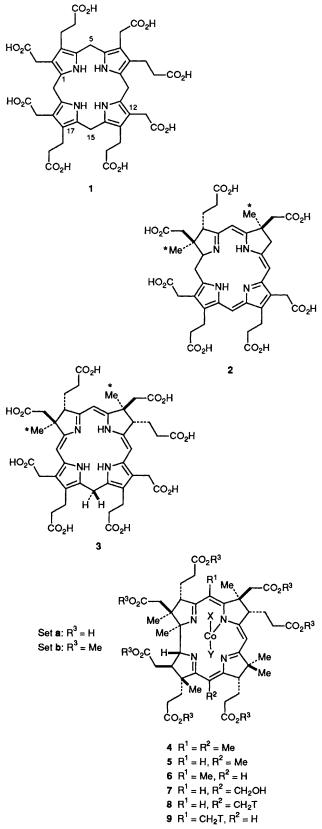
Ingeborg Grgurina, Sheetal Handa, George Weaver, Philip A. Cole and Alan R. Battersby* University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, UK

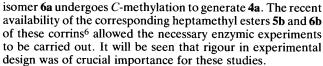
5¹-Norcobyrinic acid and 15¹-norcobyrinic acid are prepared in labelled form for enzymic experiments that show that neither one acts as a biosynthetic precursor of cobyrinic acid, a late intermediate *en route* to vitamin B_{12} .

The late stages of the biosynthetic pathway to vitamin B_{12} involve further elaboration of a well-characterised intermediate, cobyrinic acid **4a**, which in turn is built from uro'gen III **1** by a multi-step process requiring *inter alia* eight *C*-methylations.¹ The order of introduction of the first three *C*-methyl groups was known¹ by 1979 and pulse labelling² then showed that the fourth *C*-methyl group is inserted at C-17.

Extension of the pulse labelling approach³⁻⁵ indicated that the C-12 α methyl group is introduced next at C-1 and the methyls at C-5 and C-15 are the last ones with still some doubt about the order of introduction in different organisms.³⁻⁵

This knowledge still leaves open many alternative late biosynthetic sequences *en route* to cobyrinic acid **4a** but one possibility is that either 5^{1} -norcobyrinic acid **5a** nor the 15^{1} -nor





Heptamethyl 5¹-norcobyrinate⁶ **5b** was oxidised by lead tetracetate⁶ and the resultant 15¹-acetoxymethyl derivative was mildly hydrolysed to yield heptamethyl 15¹hydroxymethyl-5¹-norcobyrinate **7b**. This was reduced either by zinc and CH₃CO₂³H-tetrahydrofuran⁶ or by cyanoborotritiide and acid (conditions for the latter kindly provided by Professor Arigoni) to afford heptamethyl [15-³H-methyl]-5¹norcobyrinate **8b**. Heptamethyl [5-³H-methyl]-15¹-norcobyrinate **9b** was prepared analogously from the 15¹-nor ester⁶ **6b**.

At each stage of both preparations, HPLC conditions were developed to separate quantitatively the 5¹-nor series from the 15¹-nor series and also both nor-compounds from heptamethyl cobyrinate (cobester) itself **4b**. In this way, it was rigorously established that neither series contained any detectable quantity of the other and importantly, that both nor-series **8b** and **9b** were free of labelled cobester (as **4b**). Unlabelled samples of the 5¹-nor ester **5b** and the 15¹-nor ester **6b** were also prepared for use below and the latter was crystallised in suitable form for structure confirmation by X-ray analysis.⁷

A single preparation of the cell-free enzyme system from C. tetanomorphum⁵ was divided into three equal parts which were used in set a, Table 1, for three experiments. Experiment 1 involved incubation with $[2,7^{-14}C$ -methyl]sirohydrochlorin 2. This material 2 is known to be reduced *in situ* to precorrin-2 3, which is the true biosynthetic intermediate⁸ and thus acts as a standard substrate to check the efficacy of the enzyme system. Experiment 2, set a, involved the same amount of labelled 2 as in experiment 1 together with unlabelled 15¹-nor acid 6a prepared by hydrolysis of its ester (2 M aqueous piperidine) and in experiment 3, set a, ³H-labelled 15¹-nor acid 9a was incubated alone in the same quantity used for experiment 2. Set b, Table 1, shows the results from the equivalent set of three runs carried out with the 5¹-nor acid, 5a and 8a.

The cobyrinic acid samples isolated by ion-exchange HPLC from experiments 1 and 2 in sets a and b, Table 1, were separately esterified (5% conc. H₂SO₄ in methanol, room temp., 16-24 h), each was diluted with unlabelled cobester 4b and then multiply recrystallised. The cobester similarly isolated from experiment 3, set a was mixed with a substantial quantity of unlabelled 151-nor ester 6b and the cobester was then re-isolated using the foregoing efficient HPLC methodology. This mixing with unlabelled 151-nor ester 6b and re-isolation of the cobester by HPLC was carried out 2-3 times to ensure complete removal of the ester 6b of any residual labelled precursor from the cobester. Finally, the cobester so isolated was diluted with highly purified unlabelled cobester 4b and then multiply recrystallised. The cobester from experimental 3, set b, Table 1 was rigorously purified and crystallised in an analogous way but this time using unlabelled 5¹-nor ester **5b** at the HPLC stage.

Experiment 1, set a gave a good incorporation value, thus demonstrating the effectiveness of the enzyme preparation. Interestingly, experiment 2, set a established that the 15¹-nor acid **6a** strongly inhibits the incorporation of precorrin-2 **3** into cobyrinic acid **4a**. Finally the low level of radioactivity found in cobester **4b** from experiment 3, set a showed that the 15¹-nor acid **9a** is not a precursor of cobyrinic acid **4a**.

The experiments in set b, Table 1, also involved enzymes from *C. tetanomorphum* but from a different growth. Even though the enzyme preparation was less effective than for set a, the results from experiments 1 and 2 demonstrated that the 5¹-nor acid **5a** does not inhibit the incorporation of precorrin-2 **3** into cobyrinic acid **4a**. Experiment 3, set b suggested by comparison with experiments 1 and 2 that the 5¹-nor acid **8a** does not act as a precursor of cobyrinic acid **4a** but this result was not clear cut and required confirmation.

Table 1 Incorporation experiments

Experiment no.	Material(s) incubated ^a	Fraction (%) of original radioactivity in cobester 4b		
		Set a 15 ¹ -Nor Acid 6a and 9a	Set b 5 ¹ -Nor Acid 5a and 8a	Set c 5 ¹ -Nor Acid 5a and 8a
1	[2,7- ¹⁴ C-methyl] 2	10.5	1.7	11.9
2	[2,7-14C-methyl] 2 plus unlabelled	-0.10-	2.0	
3	nor acid The indicated [³ H-methyl] nor acid	$<0.18^{a}$ $<0.6^{a}$	$2.0 < 0.7^{b}$	<0.11 ^a

^a The quantities of 2 used in experiments 1 and 2 and of nor acid in experiments 2 and 3 were the same for each set but varied somewhat from one set to another. ^b The specific activity was still slowly falling from the penultimate to the last crystallisation; these values are, thus, maximal ones.

Accordingly, set c, Table 1 was carried out now using *Propionibacterium shermanii* cells as source of the enzyme system.⁹ The results in Table 1 made it certain that the 5¹-nor acid **8a** fails to act as a biological precursor of cobyrinic acid **4a**.

It should be noted that the levels of radioactivity present in cobester **4b** after the HPLC purification from experiment 3 of sets a, b and c, Table 1, corresponded in each case to an apparent 'incorporation' of 1-1.5%. These values only fell to those recorded after multiple recrystallisation of the cobester. Knowing that the nor-esters **5b** and **6b** and cobester **4b** are all cleanly separable by HPLC, we suspected that the labelled nor-acids **8a** and **9a** used as precursors must have contained *ca*. 1-1.5% of an epimer, which as its ester is inseparable by HPLC from cobester **4b**, but which is removable by recrystallisation. Professor Arigoni's group¹⁰ has directly proved the presence of such an epimer.

It was recognised at the outset that testing the nor-acids as possible precursors of cobyrinic acid was experimentally very demanding and by agreement with Professor Arigoni, independent studies were, therefore, made simultaneously by the two groups. Our results are in full agreement with those from the Zürich team¹⁰ and they place further limitations on the possible sequences for the late stages of biosynthesis of cobyrinic acid **4a**. Additional work is necessary to pin-point the one nature uses. Finally, these interlocking results cast doubt on an earlier report¹¹ that nor-corrins can be enzymically converted into a 5,15-dimethylated corrin.

Grateful acknowledgement is made to the Italian National Research Council and NATO for Post-doctoral Awards (to I. G.) and to The Winston Churchill Foundation for a Churchill Scholarship (to P.A.C.) and to the SERC, Roche Products Ltd. and Merck, Sharp & Dohme for financial support. We warmly thank Professor D. Argioni (Zürich) for friendly provision of information and exchange of results.

Received, 18th June 1990; Com. 0/02722G

References

- 1 Reviewed by A. R. Battersby and E. McDonald in B_{12} , ed. D. Dolphin, Wiley, New York, 1982, p. 107; F. J. Leeper, *Nat. Prod. Rep.*, 1985, **2**, 19 and 561; 1987, **4**, 441 and 1989, **6**, 171.
- 2 H. C. Uzar and A. R. Battersby, J. Chem. Soc., Chem. Commun., 1982, 1204.
- 3 H. C. Uzar and A. R. Battersby, J. Chem. Soc., Chem. Commun., 1985, 585.
- 4 A. I. Scott, N. E. Mackenzie, P. J. Santander, P. E. Fagerness, G. Müller, E. Schneider, R. Sedlmeier and G. Wörner, *Bioorg. Chem.*, 1984, 12, 356; A. I. Scott, H. J. Williams, N. J. Stolowich, P. Karuso, M. D. Gonzalez, G. Müller, K. Hlineny, E. Savvidis, E. Schneider, U. Traub-Eberhard and G. Wirth, J. Am. Chem. Soc., 1989, 111, 1897.
- 5 H. C. Uzar, A. R. Battersby, T. A. Carpenter and F. J. Leeper, J. Chem. Soc., Perkin Trans. 1, 1987, 1689.
- 6 C. Nussbaumer and D. Arigoni, Angew. Chem., Int. Ed. Engl., 1983, 22, 736.
- 7 I. Grgurina, P. R. Raithby, G. W. Sheldrick and A. R. Battersby, Acta Crystallogr., Sect. C, 1989, 45, 1589.
- 8 A. R. Battersby, K. Frobel, F. Hammerschmidt and C. Jones, J. Chem. Soc., Chem. Commun., 1982, 455.
- 9 A. R. Battersby, E. McDonald, R. Hollenstein, M. Ihara, F. Satoh and D. C. Williams, J. Chem. Soc., Perkin Trans. 1, 1977, 166.
- 10 J. Kulka, C. Nussbaumer and D. Arigoni, J. Chem. Soc., Chem. Commun., preceeding Communication.
- 11 P. Rapp and G. Ruoff, *Hoppe Seyler's Z. Physiol. Chem.*, 1973, **354**, 967.